Anti-leukemia properties of cadmium nanoparticles in the in vitro and in vivo condition: A chemobiological study

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acute myeloid leukemia, daunorubicin, Ziziphora clinopodides Lam, cadmium nanoparticles, new chemotherapeutic drug

Abstract
Introduction
The present study investigated the anti-acute myeloid leukemia effects of Ziziphora clinopodides Lam leaf aqueous extract conjugated cadmium nanoparticles.

Material and methods
To synthesize CdNPs, Z. clinopodides aqueous extract was mixed with Cd(NO3)2 .4H2O. The characterization of the biosynthesized cadmium nanoparticles was carried out using many various techniques such as UV-Vis. and FT-IR spectroscopy, XRD, FE-SEM, and EDS.

Results
The uniform spherical morphology of NPs was proved by FE-SEM images with NPs the average size of 26.78cmn. For investigating the antioxidant properties of Cd(NO3)2, Z. clinopodides, CdNPs, and Daunorubicin, the DPPH test was used. The cadmium nanoparticles inhibited half of the DPPH molecules in a concentration of 196 µg/mL. To survey the cytotoxicity and anti-acute myeloid leukemia effects of Cd(NO3)2, Z. clinopodides, CdNPs, and Daunorubicin, MTT assay was used on the human acute myeloid leukemia cell lines i.e., Murine C1498, 32D-FLT3-ITD, and Human HL-60/vcr. The IC50 of the cadmium nanoparticles was 168, 205, and 210 µg/mL against Murine C1498, 32D-FLT3-ITD, and Human HL-60/vcr cell lines, respectively. In the part of in vivo study, DMBA was used for inducing acute myeloid leukemia in mice. CdNPs similar to daunorubicin ameliorated significantly (p≤0.01) the biochemical, inflammatory, RBC, WBC, platelet, stereological, histopathological, and cellular-molecular parameters compared to the other groups.

Conclusions
As mentioned, the cadmium nanoparticles had significant anti-acute myeloid leukemia effects. After approving the above results in the clinical trial studies, these cadmium nanoparticles can be used as a chemotherapeutic drug to treat acute myeloid leukemia in humans.
Anti-leukemia properties of cadmium nanoparticles in the in vitro and in vivo condition: A chemobiological study

ABSTRACT

The present study investigated the anti-acute myeloid leukemia effects of Ziziphora clinopodides Lam leaf aqueous extract conjugated cadmium nanoparticles. To synthesize CdNPs, Z. clinopodides aqueous extract was mixed with Cd(NO$_3$)$_2$.4H$_2$O. The characterization of the biosynthesized cadmium nanoparticles was carried out using many various techniques such as UV-Vis. and FT-IR spectroscopy, XRD, FE-SEM, and EDS. The uniform spherical morphology of NPs was proved by FE-SEM images with NPs the average size of 26.78nm. For investigating the antioxidant properties of Cd(NO$_3$)$_2$, Z. clinopodides, CdNPs, and Daunorubicin, the DPPH test was used. The cadmium nanoparticles inhibited half of the DPPH molecules in a concentration of 196 µg/mL. To survey the cytotoxicity and anti-acute myeloid leukemia effects of Cd(NO$_3$)$_2$, Z. clinopodides, CdNPs, and Daunorubicin, MTT assay was used on the human acute myeloid leukemia cell lines i.e., Murine C1498, 32D-FLT3-ITD, and Human HL-60/vcr. The IC50 of the cadmium nanoparticles was 168, 205, and 210 µg/mL against Murine C1498, 32D-FLT3-ITD, and Human HL-60/vcr cell lines, respectively. In the part of in vivo study, DMBA was used for inducing acute myeloid leukemia in mice. CdNPs similar to daunorubicin ameliorated significantly ($p$$\leq$0.01) the biochemical, inflammatory, RBC, WBC, platelet, stereological, histopathological, and cellular-molecular parameters compared to the other groups. As mentioned, the cadmium nanoparticles had significant anti-acute myeloid leukemia effects. After approving the above results in the clinical trial studies, these cadmium nanoparticles can be used as a chemotherapeutic drug to treat acute myeloid leukemia in humans.

KEYWORDS Ziziphora clinopodides Lam; cadmium nanoparticles; daunorubicin; new chemotherapeutic drug; acute myeloid leukemia.

1 INTRODUCTION

Cancer is a genetic disease that includes 277 types of diseases. There are also more than 100,000 types of chemicals in our environment, of which only 35,000 have been analyzed and about 300 of them produce cancer. The remaining 65,000 chemicals in nature have not yet been tested. Cancer occurs due to uncontrolled cell division, which is the result of environmental factors and genetic disorders [1-3]. The four key genes involved in cancer cell conduction include DNA repair genes, tumor suppressor genes, oncogenes, and programmed death genes [2,3]. If a genetic mutation is produced in a cell, normal cells go out of their way and are affected by new commands that progress to cancer cells. In addition to chemicals, sunlight, shortwave, viruses and bacteria also have a special role in causing cancer [4,5]. Cancers have existed since the beginning of mankind. In recent decades, advances in computer molecular medicine have been able to not only study the causes and mechanisms of this deadly disease but also to perform better in its early diagnosis and treatment [5-7]. More than 50% of cancers are currently being treated, especially if diagnosed early. Cancer can be treated in several ways: surgery, chemotherapy, radiation therapy, immunotherapy, gene therapy, or a combination of these. Due to the relative inefficiency and very severe side effects of chemotherapy drugs, researchers and scientists have sought a new formulation of various compounds, especially metallic nanoparticles [2-5].
In recent centuries, the application of nanotechnology has played an important role in the development of science. Nanoscience has shown that if we reduce the size to nanometers, unique properties such as optical properties, electrical conductivity, hardness, and chemical reaction will be obtained. Nanoparticles are widely used because of their high surface-to-volume ratio, small size, and excellent reactivity. One of the most important advances in nanotechnology is the production and application of nanoparticles in the biological sciences [5-7]. Nanoparticles are generally effective in a wide variety of sectors that if their production is based on green chemistry, they have great applications in the fields of food, medicine, cosmetics and health. Nanoparticles centered on inorganic materials such as magnetic metals, their oxides and alloys, and semiconductors have the most studies and potential in biomedicine from diagnosis to treatment of diseases [7-9]. The effects of nanoparticles should be predictable, controllable and get the desired results with minimal toxicity. Metallic nanoparticles used in treatment and diagnosis, in addition to being non-toxic, must be biocompatible and stable in vivo. Also, by making appropriate changes in the surface of metallic nanoparticles, they will have a wide range of applications by binding to biomolecules and various carriers to cross the cell membrane and target the desired part in the body. One of the important points in the production of nanoparticles is the use of cost-effective and efficient precursors [10-12].

There are three biological, chemical, and physical methods to synthesize the nanoparticles. Chemical and physical methods are time-consuming and costly. In addition, these methods use some toxic additive chemicals that cause adverse effects on medical applications by adsorption on the surface. Applying the principles of green chemistry has decreased the use of toxic compounds or hazardous solvents, provided optimal regeneration conditions and ameliorated materials for the chemical processes, and raised new sources for green synthesis [8-10]. Therefore, one of the primary goals of green nanotechnology is to produce nanomaterials without harm to human health or environment, and to develop and design nanomaterials and products that are suitable solutions to environmental problems. The synthesis of nanoparticles by similar biological methods results in greater catalytic activity and limits the use of toxic and expensive chemicals. In biological methods, plant extracts, enzymes or proteins carrying natural resources are used to produce or stabilize nanoparticles. The nature of the materials used to make nanoparticles influences the shape, structure and morphology of these nanoparticles [7-11]. Biological systems involved in the green synthesis of nanoparticles, plants and their derivatives, as well as microorganisms such as algae, fungi, and bacteria. Plant parts such as roots, leaves, stems, fruits, and tiny parts such as the kernel and skin of the fruit are suitable to synthesize the nanoparticles because their extracts are rich in phytochemicals that act as stabilizing and reducing substances [6,8]. The use of natural plant extracts is a cheap and environmentally friendly process and does not require intermediate groups. Short time, no need for expensive equipment, precursors, high purity product and excellent quality without impurities are the features of this method. This is possible very quickly, at room temperature and pressure as well as easily on a large scale. Bio-reduction in the conversion of base metal ions is carried out by various plant metabolites such as alkaloids, phenolic compounds, terpenoids and coenzymes [9-12].

Recently, scientists have used the anti-leukemia effects of medicinal plants in several traditional medicines for synthesizing the metallic nanoparticles containing natural compounds. Hemmati et al. (2019) showed that metallic nanoparticle-chitosan composite had excellent anti-acute myeloid leukemia against Human HL-60/vcr cell line [8]. Also, Zangeneh (2019) introduced a chemotherapeutic drug formulated by metallic nanoparticles containing natural compounds to treat acute myeloid leukemia [9]. In addition to the above studies, the other studies investigated the anticancer properties of cadmium nanoparticles. In a study, Hossain and Mukherjee (2013) presented the excellent potential of cadmium nanoparticles in removing human cervical adenocarcinoma cells (HeLa) [10]. Also, in the other study, Heidari (2016) indicated the significant role of cadmium nanoparticles in combination with UV radiation to remove tissue tumors [11]. Recently, scientists have used the anti-acute leukemia effects of medicinal plants for synthesizing the metallic nanoparticles containing natural compounds. So far, the anti-acute leukemia effects of *Euphoria hirta*, *Sophora subprostrata*, *Tinospora cordifolia*, *Barleria prionitis*, *Maytenus boaria*, *Lubinus perennis*, *Cephalis acuminate*, *Solanum seaforthianum*, *Boswellia serrate*, *Phyllanthus niruri*, *Cephalotaxus harringtonia* drupacea, and *Lavendula officinalis* have been proved [12-15].

Z.* clinopodioides* aqueous extract is used in traditional medicine to treat inflammations, blood and gastrointestinal disorders, cutaneous wounds, and the common cold [15,16]. Many potentials such as anti-inflammatory [17,18], antifungal [17], antibacterial [16,18,19], and antioxidant [18,20] have been revealed as the therapeutic properties of this plant. It has chemical antioxidant components including pulegone, α- and β- pinen, terpenoids, flavonoids,
sis-isopulegone, thymol, piperitenone, and cineol [16,18,21,22]. Probably, the remedial properties of this species are related to the above compounds. Accordingly, the current study was conducted to compare the anti-acute myeloid leukemia effects of *Z. clinopodides* Lam leaf conjugated cadmium nanoparticles compared with daunorubicin by investigating the biochemical, immunological, hematological, stereological, histopathological, and cellular-molecular approaches.

2 EXPERIMENTAL

2.1 Material

Phosphate buffer solution (PBS), Sabouraud Dextrose Agar, Sabouraud Dextrose Medium, Muller Hinton Agar, Mueller Hinton Medium, carbazole reagent, 4-(Dimethylamino)benzaldehyde, Ehrlich solution, dimethyl sulfoxide (DMSO), Dulbecco's Modified Eagle Medium (DMED), hydrolysate, decampmaneh fetal bovine serum, borax-sulphuric acid mixture, 2,2-diphenyl-1-picrylhydrazyl (DPPH), and antimycotic antibiotic solution all were achieved from Sigma-Aldrich company of USA.

2.2 Synthesis of CdNPs

To obtain the aqueous extract of the plant (Figure 1), 100 gr of the dried branches of the *Z. clinopodides* leaves were poured in a container containing 1000 mL boiled water, and the container lid was tightly closed for 4 h. Then, the container’s content was filtered, and the remaining liquid was placed on a bain-marie to evaporate. Finally, a tar-like material was obtained, which was powdered by a freeze dryer [23].

A 20 mL of the extract (100 µg/mL) was added to 20 mL of Cd(NO$_3$)$_2$.4H$_2$O (0.15 mM). The reaction mixture was stirred for 12 h at room temperature. The appearance of a yellow colloidal solution approved the formation of cadmium nanoparticles. The obtained CdNPs were washed three times with water and centrifuged at 10000 rpm for 15 min. The common techniques of organic chemistry, i.e. FT-IR and UV-Vis. spectroscopy, SEM, XRD, and EDS were used to characterize of CdNPs [23].

2.3 DPPH test

Free radicals are unstable atoms that have one or more unpaired electrons. These active species are very harmful due to their high reactivity. They are most often formed when oxygen molecules in the body split into separate unstable atoms. This process can turn into a chain reaction. Free radicals’ excessive production in the body causes cell damage and oxidative stress. Genetics and the environment affect the extent of free radical damage in individuals. These active molecules are produced as part of the body's natural biological processes. One of the most important free radicals is DPPH. DPPH is widely used to study the antioxidant activities of natural compounds and nanoparticles [24].

In this method, the antioxidant activity of nanoparticles is measured for DPPH radical scavenging. The basis of the action is the reduction of the alcoholic solution of DPPH in the presence of hydrogen-giving antioxidants, especially phenolic compounds. To achieve the IC50 of the samples, 11 different concentrations of nanoparticles were prepared and the percentage of inhibitory versus concentration was used to plot. In practice, 300 µl of 1 M DPPH was combined with 100 µl of diluted sample to a final volume of 2 ml using methanol. After half an hour in the dark, the absorbance was read at 517 nm and the inhibitory percentage was obtained using the following formula:

$$\text{Inhibition (\%)} = \frac{\text{Sample A}}{\text{Control A}} \times 100$$

In this formula, “Control A” shows the negative control of light absorption that lacks nanoparticles, and “Sample A” expresses the amount of light absorption of different concentrations of nanoparticles. [24]
2.4 Analysis of cytotoxicity of CdNPs
In this experiment, Cd(NO$_3$)$_2$, Z. clinopodides, CdNPs, and Daunorubicin were examined for their cytotoxicity activities against HUVEC, Murine C1498, 32D-FLT3-ITD, and Human HL-60/vcr cell lines using the common cytotoxicity test i.e., MTT assay in vitro condition.

Because nanoparticles are not soluble directly in 1640-RPMI medium and also the solvent of dimethyl sulfoxide nanoparticles (DMSO) itself has cytotoxic effects, to eliminate the effect of this substance on treated cells, its amount in the final solution is considered less than 1%. Dimethyl sulfoxide is not toxic to concentrations less than 1% and the concentration of this solvent is important in this regard. For this purpose, 1000 μg of nanoparticle was dissolved in 100 μl of dimethyl sulfoxide solvent after weighing. Then 1 ml of culture medium was added for better dissolution and finally the volume of solution was increased to 24 ml using culture medium: Then, successive dilutions of this stock were used in the proportions of 1-1000 μg/ml. Eleven concentrations were used for the cell lines [25].

In this study, 100 microliters of culture medium containing $10^4$ cells per plate 96 were placed. After 24 hours, incubation of concentrations of 1-1000 μg/ml of nanoparticles was added to the cells, and incubated for 24, 48, and 72 hours, respectively. After these times, 20 μl of MTT plate with a concentration of 5 mg/ml was added to each cell and incubated in the dark for another 4 hours. After some time, the MTT medium was carefully removed, and 200 microliters of acidified isopropanol were added to each plate to remove the purple formanes. After 15 minutes of incubation at room temperature, the light absorption of each well was read using an ELISA at 570 nm against a reference wavelength of 690 nm. The findings were reported as cell survival and IC50 (concentration that inhibits cell growth up to 50%) based on the concentration curve (μg/ml) [25].

It should be noted that the effect of each concentration of the nanoparticles on cell lines was investigated in five independent experiments. According to the values of light absorption obtained by the ELISA reader, the percentage of growth inhibition related to each concentration was calculated using the following formula:

$$\text{Cell viability (\%)} = \left(\frac{\text{Sample A}}{\text{Control A}}\right) \times 100$$

Finally, linear regression was done to gain IC50, which indicates the nanoparticles concentration, which causes 50% cancer cell growth inhibition. Using the curve, the line equation for cancer cells was obtained, respectively, then by replacing 50% inhibition in the equation, the IC50 value for cancer cells was obtained [25].

2.5 In vivo design
At the beginning of the study, Acute myeloid leukemia induction was done with 40 mg/kg body weight intravenous injection of 7,12-Dimethylbenz[a]anthracene (DMBA) biweekly for 6 weeks in 75 animals. One group was considered as a healthy control and injected with normal saline (n=15). After 6 weeks, the 75 DMBA-treated mice were randomly divided into five subgroups, including untreated receiving distilled water, and four groups receiving 1mg/kg body weight intravenous injection of Cd(NO$_3$)$_2$, Z. clinopodides, CdNPs, and Daunorubicin separately. During this period, the healthy control group received normal saline. The mice underwent injection treatment for 24 days. Dosing was repeated for a total of 6 cycles of dosing. The mice were weighed seven times during the study. At the end of day 24 of treatment after anesthetizing the animals, the blood samples were drawn immediately from the animals’ hearts. Also, the mice livers and spleens were harvested to analyze biochemical and stereological factors [26].

In serological parameters, the volumes of hepatocyte, central vein, portal vein, sinusoid, bile duct, and hepatic artery in the liver and the volumes of white pulp, red pulp, vessels, marginal zone, and follicle in the spleen were investigated [13].

Biochemical (SOD, CAT, GR, GPx, ALP, AST, ALT, GGT, cholesterol, LDL, HDL, triglyceride, total protein, albumin, total and conjugated bilirubin, glucose, urea, creatinine, ferrous, ferritin, and erythropoietin), immunological (Pro-inflammatory (IL1, IL6, IL12, IL18, IFNY, and TNFα) and anti-inflammatory (IL4, IL5, IL10, IL13, and IFNα) cytokines) parameters concentration were measured in the serum by ELISA procedure and determined using kits (Ziest Chem Diagnostics) according to the manufacturers’ instructions [27,28]. The number of total WBC, blast, lymphocyte, monocyte, neutrophil, eosinophil, basophil, platelet, RBC, and nRBC, and the
level of HB, PCV, MCV, MCH, and MCHC were analyzed in the plasma by automatic hematology analyzer (Sysmex XS 800i) [26].

In the cellular and molecular part of this study, the mRNA expression of S1PR1 or Sphingosine-1-phosphate receptor-1 and S1PR5 or Sphingosine-1-phosphate receptor-5 was measured by Real-Time PCR in blood lymphocytes [26].

All stereological, biochemical, immunological, hematological, and cellular and molecular analyses were done in duplicate.

2.6 Statistical analysis

SPSS statistical software version 22 was used for data analysis and the findings were determined as the mean standard deviation of 5 replications. Data were analyzed using one-way analysis of variance and Duncan post hoc test and the significance level in the test was considered 0.05.

3 RESULTS AND DISCUSSION

Recently, the anti-angiogenic and anti-cancer properties of metallic nanoparticles have been determined and the findings have revealed that metallic nanoparticles can be used as a unique anti-cancer supplement. But the organic solvents used to produce these nanoparticles are toxic and can have devastating environmental effects. Hence there is a great tendency to use healthy methods for synthesizing metallic nanoparticles. Recently, studies have begun under the title Green Chemistry, which seeks an environmentally friendly way for nanoparticles synthesis [1-3]. In recent years, metallic nanoparticles have been synthesized extracellularly using various plant and microbial extracts. In addition, many studies are underway to investigate the antibacterial, antioxidant and anti-tumor properties of these natural products [12-14]. In this regard, a study assessed the antibacterial activities of metallic nanoparticles synthesized biologically. Another study evaluated the anticancer, antibacterial, and antioxidant properties of metallic nanoparticles yielded by plant extract [5-10]. The anti-inflammatory, antioxidant and antimicrobial effects of metallic nanoparticles green-synthesized by medicinal plants are well known [12-14]. It also affects a range of molecular targets and signaling pathways, such as NF-JB, AKT/mTOR, and HIF-1A, and as a result, it plays a role in inhibiting cancer cell proliferation, metastasis, angiogenesis and also inducing apoptosis. Previous studies have also shown that metallic nanoparticles containing medicinal plants cause deformity and perforation of cancer cells, resulting in their death [10-14].

3.1 In vitro anti-acute myeloid leukemia property of CdNPs

Investigation of cell proliferation and survival is one of the most important and basic techniques in cell laboratories. This study requires accurate quantification of the number of living cells in the cell culture medium. Therefore, cell survival calculation methods are necessary to optimize cell culture conditions, evaluate cell growth factors, detect antibiotics and anticancer drugs, evaluate the toxic effects of environmental pollutants, and study apoptosis. Many methods can be used for such purposes, but indirect methods using fluorescent or dye (chromogenic) markers provide very fast large-scale methods. Among these methods, measurement of cell survival by MTT method (3-(4,5-dimethylthiazol-2-yl) -2,5-diphenyltetrazolium bromide) is the most widely used method. This method is a colorimetric method to study cell proliferation and survival, introduced in 1983 by Mossman. The method basis is based on mitochondrial activity. Mitochondrial activity in living cells is stable and therefore a raise or reduce in the living cells number is linearly related to mitochondrial activity [25-27]. MTT tetrazolium dye is revived in active cells. Mitochondrial dehydrogenases in living cells break the tetrazolium ring and yield NADPH and NADH, leading to forming a purple insoluble deposit called formazan. This precipitate can be dissolved by dimethyl sulfoxide or isopropanol. On the other hand, dead cells do not have this ability and therefore do not reveal a signal. Dye formation is used as a marker of living cells. The color intensity produced is measured at a 540 to 630 nm wavelength and is directly proportional to the living cells number. High safety and providing a colorimetric and non-radioactive system are important advantages of this method. This kit is very
easy to use, has high sensitivity and accuracy and can detect less than 950 cells. On the other hand, it has high efficiency for measuring cell proliferation, survival and mortality, and its implementation method does not require time-consuming washing steps and transfer from one plate to another. Examples studied in this method are adhesive or suspended cells and proliferating or non-proliferating cells [25-29].

In the current experiment, the treated cells with many concentrations of the present Cd(NO$_3$)$_2$, Z. clinopodides, CdNPs, and Daunorubicin were examined by MTT test for 48h regarding the cytotoxicity property on normal (HUVEC) and acute myeloid leukemia (32D-FLT3-ITD, Murine C1498, and Human HL-60/vcr) cell lines (Figure 2). The absorbance rate was determined at 570nm, which indicated extraordinary viability on normal cell line (HUVEC) even up to 1000μg/mL for Cd(NO$_3$)$_2$, Z. clinopodides, CdNPs, and Daunorubicin. In the case of acute myeloid leukemia cell lines, the viability of them reduced dose-dependently in the presence of Cd(NO$_3$)$_2$, Z. clinopodides, CdNPs, and Daunorubicin. The IC50 of Cd(NO$_3$)$_2$, Z. clinopodides, CdNPs, and Daunorubicin against Murine C1498 cell line were 735, 380, 197, and 168μg/mL, respectively, against 32D-FLT3-ITD cell line were 985, 500, 246, and 205 μg/mL, respectively. About the Human HL-60/vcr, the IC50 of Z. clinopodides, CdNPs, and Daunorubicin were 641, 342, and 210, respectively. The size of the nanoparticles affect their anticancer properties, so that the nanoparticles with the size lower than 100 nm have unique anticancer potentials [29].

![Figure 2](image)

### 3.2 Antioxidant property of CdNPs against DPPH

Oxidation is the electrons transfer from an atom and is the aerobic life and metabolism part of living organisms. Oxygen is the receptor for electrons in the electron transport system, which yields energy from ATP (Adenosine triphosphate) in the body. Under certain conditions, oxygen may become a single electron and release free radicals. When oxygen becomes a single electron, it is called reactive oxygen species (ROS). Oxidative loss to proteins, DNA, and other macromolecules is one of the internal causes of degenerative diseases such as aging, cardiovascular disease, cancer, immune system deficiency, cataracts, and abnormal brain function. Single oxygen, high-energy, mutagenic oxygen, can be produced by lipid peroxidation by the transmission of energy from light or the respiratory tract of neutrophils [30,31]. Some free radicals have positive roles such as regulating cell growth, phagocytosis, energy production, intracellular signals, or the important biological compounds synthesis. Antioxidants produced in the body fight free radicals with two systems: enzymatic defense and non-enzymatic defense. Superoxide dismutase, catalase, and glutathione peroxidase metabolize lipid peroxide, hydrogen peroxide, and superoxide and prevent the production of toxic hydroxyl radicals [32,33]. In non-enzymatic defense, there are two classes of fat-soluble antioxidants (such as carotenoids and vitamin E) and water-soluble (glutathione and vitamin C) that trap free radicals. These two systems help neutralize oxidants. However, oxidants can escape from antioxidants and damage tissues. In this case, the activated antioxidant repair system (which is the enzymes lipase, protease, transferase and DNA repair enzymes), counteract the oxidant effects. However, due to deficiencies in the production of antioxidants in the body or due to physiopathological factors and situations (such as smoking, air pollution, UV radiation, diets containing high unsaturated fatty acids, inflammation, ischemia, bleeding, etc) that ROS are yielded in large quantities at the wrong place and time, oral antioxidants are needed to counteract the oxidative damage cumulative effects [30-33].

In our research, Z. clinopodides leaf aqueous extract, Daunorubicin, and CdNPs similar to BHT showed a notable concentration-dependent DPPH radical scavenging effect. The IC50 values of Z. clinopodides leaf aqueous extract, BHT, Daunorubicin, and CdNPs were 428, 414, 242, and 196, respectively (Figure 3).

The interaction between the Z. clinopodides leaf aqueous extract, Daunorubicin, and CdNPs and DPPH might have occurred through the transfer of electrons and hydrogen ions to 2,2-diphenyl-1-picrylhydrazyl radical to form a stable 2,2-diphenyl-1-picrylhydrazine molecule (DPPH) [30-34].
3.3 *In vivo* anti-acute myeloid leukemia properties of CdNPs

To induce the acute myeloid leukemia, DMBA was used. DMBA causes many problems in the body such as self-immune, hepatic, splenic, vascular, central nervous system, respiratory, and renal disorders and acute myeloid leukemia [35]. In details, DMBA-induced acute myeloid leukemia reduces the platelet, lymphocyte, and RBC counts and increase the WBC, blast, eosinophil, neutrophil, monocyte, and basophil [26,36,37,38].

As is observed in Figure 4, the anti-inflammatory cytokines including IFNα, IL4, IL5, IL10, and IL13 reduced significantly (p≤0.01) and the pro-inflammatory cytokines including IFNY, TNFα, IL1, IL6, IL12, and IL18 enhanced significantly (p≤0.01) in the DMBA-treated group. Cd(NO\textsubscript{3})\textsubscript{2}, *Z. clinopodides* leaf aqueous extract, CdNPs, and Daunorubicin significantly (p≤0.01) ameliorated the pro-inflammatory and anti-inflammatory cytokines as compared to the untreated group. The best results have been seen in the CdNPs-treated group.

In the present study, the number of total WBC, blasts, monocyte, neutrophil, eosinophil, basophil, and nRBC counts raised significantly (p≤0.01) and the number of RBC, platelet, and lymphocyte counts and the other RBC parameters including Hb, PCV, MCV, MCH, and MCHC decreased significantly (p≤0.01) in the DMBA-treated group (Figures 5,6). The above results show that DMBA successfully induced acute myeloid leukemia in mice. Cd(NO\textsubscript{3})\textsubscript{2}, *Z. clinopodides* leaf aqueous extract, CdNPs, and Daunorubicin ameliorated significantly (p≤0.01) the hematological parameters. The best results have been seen in CdNPs and Daunorubicin-treated group. In the present study, the anti-hemolytic activity of Cd(NO\textsubscript{3})\textsubscript{2}, *Z. clinopodides* leaf aqueous extract, CdNPs, and Daunorubicin may be associated with their antioxidant potentials. Many studies similar to our study revealed significant antioxidant effects of cadmium nanoparticles [10,11, 39-41].

In our study, mRNA expression of S1PR1 in lymphocytes increased by 33.2, 8.1, 14.7, 20, and 23.1 folds in control, Cd(NO\textsubscript{3})\textsubscript{2}, *Z. clinopodides* leaf aqueous extract, CdNPs, and Daunorubicin, respectively compared to the untreated group. Also, the findings of S1PR5 mRNA expression revealed that the number of folds increased 36.8, 12.1, 17.3, 24.2, and 26.9 in control, Cd(NO\textsubscript{3})\textsubscript{2}, *Z. clinopodides* leaf aqueous extract, CdNPs, and Daunorubicin, respectively in comparison to the untreated group. The best results were seen in the Daunorubicin and CdNPs-treated groups (Figure 7).

In this study, DMBA-induced Acute myeloid leukemia significantly (p≤0.01) increased the concentrations of ALP, AST, ALT, GGT, cholesterol, LDL, triglyceride, total and conjugated bilirubin, glucose, urea, creatinine, ferrous, ferritin, and erythropoietin and decreased HDL, total protein and albumin in serum. Cd(NO\textsubscript{3})\textsubscript{2}, *Z. clinopodides* leaf aqueous extract, CdNPs, and Daunorubicin significantly (p≤0.01) ameliorated the above parameters. There were no significant changes (p≤0.01) between CdNPs, Daunorubicin, and control groups in GGT, triglyceride, glucose, and creatinine levels. The best results in improving the biochemical parameters were seen in the CdNPs-treated group (Table).
In the recent study, the levels of the liver, spleen, and serum SOD, CAT, and GPx increased and GR reduced significantly \((p \leq 0.01)\) at \(\text{Cd(NO}_3\text{)}_2\), \textit{Z. clinopodides} leaf aqueous extract, CdNPs, and Daunorubicin-treated mice in comparison to the untreated group. Agreement with our study, in the study of Saravanan and Ponmurugan (2013), it revealed the strong antioxidant effect of medicinal plants with ameliorating the concentrations of SOD, CAT, and GPx [42].

Table

In the present study, the body weight of the untreated group decreased \((p \leq 0.01)\) after six weeks of acute myeloid leukemia induction.

Cancer cachexia is a wasting syndrome involving loss of muscle mass and fat directly caused by tumor factors [43-47]. But, the administration of \(\text{Cd(NO}_3\text{)}_2\), \textit{Z. clinopodides} leaf aqueous extract, CdNPs, and Daunorubicin raised \((p \leq 0.01)\) the body weight in comparison to the untreated group (Figure 8).

Figure 8

The results analysis of Figures 9-11 indicated that DMBA significantly \((p \leq 0.01)\) increase the leukemic myeloblasts infiltrating in the spleen and liver. With this infiltrating, the volumes and weights of spleen and liver and their sub-compartments raised in the untreated as compared to other groups.

In detail, DMBA enhanced significantly \((p \leq 0.01)\) the volumes of hepatocyte, central vein, portal vein, sinusoid, bile duct, and hepatic artery in the liver and the volumes of white pulp, marginal zone, and follicle in the spleen. Treatment with \(\text{Cd(NO}_3\text{)}_2\), \textit{Z. clinopodides} leaf aqueous extract, CdNPs, and Daunorubicin reduced significantly \((p \leq 0.01)\) the leukemic myeloblasts infiltrating in the liver and spleen. The best results were observed in the CdNPs and Daunorubicin-treated groups.

Figure 9
Figure 10
Figure 11

3.4 Characterization of CdNPs

Using EDS (see Figure 12), the elemental composition profile of the biosynthetic CdNPs showed the presence of cadmium biosynthesized NPs considering the signals appearing at 3.16 keV and 3.34 keV corresponded to the L\(\alpha\) and L\(\beta\) peaks of cadmium while the peaks at around 2.3-2.5 keV corresponded to the K\(\alpha\) and K\(\beta\) peaks of sulfur. These signals are as well as match to a previous study on the synthesized CdsNPs using \textit{C. reinhardtii} extract[48]; The other signals including Ok\(\alpha\) and CK\(\alpha\) belong to the organic molecules present in \textit{Z. clinopodioides} extract that linked to the surface of CdNPs.

Figure 12

Fig. 13 shows the XRD pattern of the biosynthesized CdNPs which approve the synthesis of cadmium sulfide. The peaks at 2\(\theta\) values of 26.68, 44.16, 52.47, and 70.65 are indexed as (111), (202), (311), and (313) planes which are in good agreement with those of clinic phase CdS obtained from standard database ICDD PDF card no. 96-900-0068. The crystallinity of the sample has been accepted using the XRD diffractogram. The purity of the sample is approved because of the absence of unmatched peaks. The peaks at different degrees are also reported previously [48-55].

The CdNPs nanoparticles had an average crystal size of 25.65 nm that calculated using X Scherrer equation.

\[
D = \frac{k\lambda}{\beta \cos\theta}
\]
In our literature review, cadmium sulfide and cadmium oxide are the common products of the biosynthesis of cadmium nanoparticles. The researches have reported various crystal sizes for the biosynthetic CdSNPs or CdONPs according to the XRD patterns. Instead of, for CdSNPs the crystal size of 3.42 nm has been reported for the synthetic CdSNPs using *Annona muricata* leaf extract [23]; 6 nm for *Chlamydomonas reinhardtii* [48]; 1.47 nm for Banana peel extract [49]; 18 nm and 12 nm for *Calotropis gigantea* leaf extract [51]; 5.67 nm for *Mimosa pudica* [52-55]; 4 nm for *Murraya koenigii* extract [56]. On the other hand, 10.86 nm have been reported for the synthetic CdONPs using *Limonia acidissima* leaves extract [57]; 18.5-61.9 nm for *Abelmoschus esculentus* [58]; 22 nm for *Andrographis paniculata* leaf extract [59]; 25 to 50 nm for *Agathosma betulina* [60]; 22.9 and 35.0 nm for *Hibiscus sabdariffa* flower extract [61]; 44-56 nm for *Aloe barbadensis* [62]; 28.92 to 44.95 nm for *Parkia speciosa* seed extract [63].

Figure 13

FT-IR (Fourier Transform Infrared) has been a suitable technique for analyzing materials in the laboratory. An infrared spectrum represents the fingerprint of the sample under test with absorption peaks, which depends on our vibrational frequencies between the atomic bonds of that material. Since each substance has its own atomic bonds, no two compounds with the same infrared spectrum are alike. Hence, infrared spectroscopy can be effective in better identification (qualitative analysis) of different types of materials. Also, the peak sizes are in the range indicating the amount of material present. Advanced software algorithms make this spectroscopy a great tool for quantitative analysis [48,51,52].

In FT-IR spectra of metal oxides, the vibration band for metal-oxygen bond usually appears in 500 to 700 cm\(^{-1}\). Fig.14 presents the spectra of CdNPs. The peaks at 524, 621, and 678 cm\(^{-1}\) are attributed to the bending vibration of Cd-S. These peaks for biosynthetic CdNPs have been reported previously with a small difference in the respective wavenumber [48,51,52]. FT-IR spectroscopy is a reliable strategy to evaluate the secondary plant metabolites as the capping and reducing agents of Cd(NO\(_3\))\(_2\) precursor to CdNPs. According to the findings of this report, the FT-IR spectra of CdNPs and *Z. clinopodioides* extract were very similar to each other confirming the successful biosynthesis of the cadmium nanoparticles. The presence of different-IR bands correlates with the existence of various functional groups in *Z. clinopodioides* extract. For instance, the peaks in 3371 and 2939 cm\(^{-1}\) for O-H and aliphatic C-H stretching, the bands at a range of 1421 to 1749 cm\(^{-1}\) for C=C and C=O stretching, and the peaks at 1261 and 1081 cm\(^{-1}\) for -C-O and -C-O-C stretching are generally considered to confirm the presence of various valuable natural compounds such as phenolic, flavonoid, and carboxylic compounds as secondary metabolites in *Z. clinopodioides* which have been reported previously [53,54]. These compounds bind to metallic NPs via hydroxyl and carbonyl functional groups as the main stimulus to reduce, stabilize, and disperse of metallic nanoparticles [55].

Figure 14

The FE-SEM device is one of the most powerful tools used in various fields, including nanotechnology, which uses electron bombardment to produce images of objects as small as 10 nanometers. The bombardment of the sample causes positively charged electrons to be released from the sample to the plate, where these electrons become signals [60-64].

FE-SEM images of CdNPs shown in Fig. 15 (a-d) depict a spherical morphology for the prepared CdNPs reported previously [64]. The figure also confirms the uniformity, well-dispersed, and homogeneity of the CdNPs. Similar to the other metallic nanoparticles being synthesized using green chemistry approaches, a tendency to aggregate is observed for CdNPs. This property has been reported for other types of biosynthesized nanoparticles, viz. CdNPs, CuNPs, AgNPs, and TiNPs [48,55,65,66]. On average, the diameter of particle size for CdNPs was 26.78
nm. The size is very close to the obtained one from the XRD pattern. In our review of literature, different sizes have been reported for the biosynthesized CdSNPs or CdONPs using herbal extracts. For example, 3 to 5 nm has been reported for synthetic CdSNPs using tea leaf extract [64]; 4 nm for A. muricata [23]; 2-7 nm for C. reinhardtii [48]; 1.48 nm for banana peel extract [49]; 5-20 nm for Fungus of Fusarium oxysporum [50]; 20 nm for C. gigantea leaf extract [51]; and 20 nm for M. Koenigii [56]. On the contrary, bigger sizes have been reported for biosynthetic CdONPs: 10-30 nm for Limonia acidissima leaves extract [57]; 78 nm for A. esculentus fruit extract [58]; 164 nm for A. paniculata leaf extract [59]; 100 nm for A. betulina [60]; 113 nm for H. Sabdariffa flower extract [61]; 97-129 nm for A. Barbadensis [62]; 20-50 nm for P. speciosa seed extract [63]; and 36-57 nm for Leucaena leucocephala [67].

Figure 15

UV-Vis is based on the irradiation of ultraviolet and visible photons on the sample and measures the rate of passage or absorption of matter at different wavelengths in the range of 200 to 1100 nm. It is possible to measure the spectrum for samples in solution, solid as well as thin layers. The size of solid samples should be larger than 20 mm. This test is not possible for powder samples. One of the important applications of UV device is to determine the concentration of the unknown solution. By having the original sample and its solvent and making several solutions with different percentages and drawing a calibration diagram based on the calculation of the maximum land, the concentration of the unknown solutions can be calculated [50-53].

The UV-Vis. spectrum of biosynthesized CdNPs is shown in Figure 16. The surface plasmon resonance of CdNPs was confirmed by UV-Vis. and compared with a previous report, the appearance of three bands at 393, 450, and 578 nm approved the formation of CdNPs. However, in comparison with bulk absorption, a blue shift is observed in the spectrum. According to the previous studies, the decrease in the particle size is the main reason for this observation [48,49]. This band is also characteristic of CdS particles in the quantum size range [50].

Figure 16

4 CONCLUSION
In the present experiment, Z. clinopodides extract-based cadmium nanoparticles (CdNPs) were successfully synthesized. The NPs has been completely characterized using UV-Vis. and FT-IR spectroscopy; XRD, FE-SEM, and EDS. In vivo field, CdNPs significantly (p≤0.01) regulated the levels of biochemical, immunological, hematological, histopathological, and cellular-molecular parameters similar to daunorubicin. In vitro field, CdNPs had low cell viability dose-dependently against Human HL-60/vcr, 32D-FLT3-ITD, and Murine C1498 cell lines without any cytotoxicity on HUVEC cell line. According to the above results, it appears that CdNPs may be applied to treat acute myeloid leukemia in humans.

REFERENCES
<table>
<thead>
<tr>
<th>Parameters</th>
<th>Groups (n=10)</th>
</tr>
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<tbody>
<tr>
<td></td>
<td>Control</td>
</tr>
<tr>
<td>Serum ALP (IU/L)</td>
<td>176.3±10.1\textsuperscript{a}</td>
</tr>
<tr>
<td>AST (IU/L)</td>
<td>99.6±8.3\textsuperscript{b}</td>
</tr>
<tr>
<td>ALT (IU/L)</td>
<td>31.5±2.5\textsuperscript{a}</td>
</tr>
<tr>
<td>GGT (IU/L)</td>
<td>7.2±0.8\textsuperscript{a}</td>
</tr>
<tr>
<td>Cholesterol (mmol/L)</td>
<td>4.8±0.4\textsuperscript{a}</td>
</tr>
<tr>
<td>LDL (mmol/L)</td>
<td>3.2±0.2\textsuperscript{a}</td>
</tr>
<tr>
<td>HDL (mmol/L)</td>
<td>5.9±0.5\textsuperscript{a}</td>
</tr>
<tr>
<td>Triglyceride (mmol/L)</td>
<td>0.6±0.0\textsuperscript{a}</td>
</tr>
<tr>
<td>Total protein (g/dL)</td>
<td>8.1±0.9\textsuperscript{a}</td>
</tr>
<tr>
<td>Albumin (g/dL)</td>
<td>3.2±0.2\textsuperscript{a}</td>
</tr>
<tr>
<td>Total bilirubin (mg/dL)</td>
<td>0.41±0.0\textsuperscript{a}</td>
</tr>
</tbody>
</table>

**TABLE** The levels of biochemical parameters in tested groups (p≤0.01).
<table>
<thead>
<tr>
<th>Parameter</th>
<th>Value</th>
</tr>
</thead>
<tbody>
<tr>
<td>Conjugated bilirubin (mg/dL)</td>
<td>0.09±0.0a, 0.23±0.0a, 0.15±0.0b, 0.16±0.0b, 0.14±0.0b, 0.10±0.0a</td>
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<tr>
<td>Glucose (g/dL)</td>
<td>76.7±7.5a, 124.3±10.7b, 85.6±9.2a, 88.7±10a, 84.2±8.3a, 80.3±8.6a</td>
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<tr>
<td>Urea (mg/dL)</td>
<td>12.1±0.9a, 35.4±2.6d, 21.4±1b, 26.4±1.3c, 21.7±0.9b, 19.7±0.8b</td>
</tr>
<tr>
<td>Creatinine (mg/dL)</td>
<td>0.6±0.0a, 1.9±0.0b, 0.8±0.0a, 1±0.0a, 0.8±0.0a, 0.8±0.0a</td>
</tr>
<tr>
<td>Ferrous (µg/dL)</td>
<td>196.5±11.5a, 395.6±14.3d, 256.6±12.5c, 315.3±13.2c, 300.2±12c, 243.5±9.6b</td>
</tr>
<tr>
<td>Ferritin (µg/g protein)</td>
<td>165.7±11.4a, 419.6±15.2a, 285.3±11.7b, 343.2±12.9c, 327.6±11.2c, 259.5±12.4b</td>
</tr>
<tr>
<td>Erythropoietin (mU/mL)</td>
<td>9.6±0.8a, 18.7±1.4a, 43.7±3.2a, 26.5±2b, 30.5±1.6c, 26.8±1.4b</td>
</tr>
<tr>
<td>Creatinine (mg/dL)</td>
<td>0.6±0.0a, 1.9±0.0b, 0.8±0.0a, 1±0.0a, 1±0.0a, 0.8±0.0a</td>
</tr>
<tr>
<td>Urea (mg/dL)</td>
<td>12.1±0.9a, 35.4±2.6d, 21.4±1b, 26.4±1.3c, 21.7±0.9b, 19.7±0.8b</td>
</tr>
<tr>
<td>Creatinine (mg/dL)</td>
<td>0.6±0.0a, 1.9±0.0b, 0.8±0.0a, 1±0.0a, 0.8±0.0a, 0.8±0.0a</td>
</tr>
<tr>
<td>Ferrous (µg/dL)</td>
<td>196.5±11.5a, 395.6±14.3d, 256.6±12.5c, 315.3±13.2c, 300.2±12c, 243.5±9.6b</td>
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</tr>
<tr>
<td>Erythropoietin (mU/mL)</td>
<td>9.6±0.8a, 18.7±1.4a, 43.7±3.2a, 26.5±2b, 30.5±1.6c, 26.8±1.4b</td>
</tr>
<tr>
<td>Spleen SOD (µg/mg protein)</td>
<td>9.6±0.8a, 3.1±0.3a, 5.4±0.4a, 1.8±0.1a, 31.6±2.9a, 17.8±1.5a</td>
</tr>
<tr>
<td>CAT (µ/mg protein)</td>
<td>9.6±0.8a, 3.1±0.3a, 5.4±0.4a, 1.8±0.1a, 31.6±2.9a, 17.8±1.5a</td>
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<tr>
<td>Spleen GR (U/ml protein)</td>
<td>5.4±0.4a, 14.6±1.2c, 8±0.7c, 10.1±0.7d, 6.3±0.5c, 7.3±0.7b</td>
</tr>
<tr>
<td>Spleen GPx (U/ml protein)</td>
<td>5.4±0.4a, 14.6±1.2c, 8±0.7c, 10.1±0.7d, 6.3±0.5c, 7.3±0.7b</td>
</tr>
<tr>
<td>Liver SOD (µg/mg protein)</td>
<td>44.2±3.6a, 17.8±1.5a, 3.8±0.3a, 1.7±0.1a, 44.2±3.6a, 25.8±1.4a</td>
</tr>
<tr>
<td>Liver CAT (µ/mg protein)</td>
<td>19.8±1.2d, 36.2±2.3b, 15.7±0.9d, 3.0±0.0b, 19.8±1.2d, 8.9±0.7d</td>
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<tr>
<td>Liver GR (U/ml protein)</td>
<td>19.8±1.2d, 36.2±2.3b, 15.7±0.9d, 3.0±0.0b, 19.8±1.2d, 8.9±0.7d</td>
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<td>Liver GPx (U/ml protein)</td>
<td>19.8±1.2d, 36.2±2.3b, 15.7±0.9d, 3.0±0.0b, 19.8±1.2d, 8.9±0.7d</td>
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**FIGURE 1** The image of *Ziziphora clinopodides* Lam.

**FIGURE 2** Percent viability measured on HUVEC, 32D-FLT3-ITD, Human HL-60/vcr, and Murine C1498 after treatment with present Cd(NO₃)₂, *Z. clinopodides* leaf aqueous extract, CdNPs, and Daunorubicin.

**FIGURE 3** Antioxidant activity of Cd(NO₃)₂, *Z. clinopodides* leaf aqueous extract, CdNPs, Daunorubicin, and BHT.

**FIGURE 4** The levels of immunological parameters in tested groups (p≤0.01).
A: Anti-inflammatory cytokines.
B: Pro-inflammatory cytokines.

**FIGURE 5** The number of platelet, RBC, nRBC, and WBC in the experimental groups (p≤0.01).

**FIGURE 6** The levels of RBC parameters in tested groups (p≤0.01).
FIGURE 7 Expression fold change of S1PR1 and S1PR5 in control, Cd(NO$_3$)$_2$, *Z. clinopodides* leaf aqueous extract, CdNPs, and Daunorubicin groups in comparison to the untreated group.

FIGURE 8 The weights of the body in tested groups.

FIGURE 9 The percentage of leukemic myeloblasts infiltrating to spleen and liver in tested groups ($p \leq 0.01$).

FIGURE 10 The weights and volumes of liver and spleen in tested groups ($p \leq 0.01$).

FIGURE 11 The levels of stereological parameters of liver and spleen in tested groups ($p \leq 0.01$).
A: Spleen.
B: Liver.

FIGURE 12 EDS pattern of CdNPs.

FIGURE 13 XRD data of CdNPs.

FIGURE 14 FT-IR spectra of CdNPs.

FIGURE 15 FE-SEM image of CdNPs.

FIGURE 16 UV–vis absorption spectra of CdNPs.
FIGURE 2
FIGURE 4

A

B
FIGURE 5

Graphs showing the effect of different treatments on various blood cell counts. The treatments include Control, Untreated, Cd(NO3)2, Z. clinopodioides, CdNPs, and Daunorubicin. The cell types include RBC, nRBC, Platelet, WBC, Neutrophil, Blast, Monocyte, Lymphocytes, Eosinophils, and Basophils. The graphs display the cell counts in $\times 10^7/L$.
FIGURE 6

Bar charts showing the comparison of MCV, PCV, MCHC, and Hb across different treatments: Control, Cd(NO3)2, Z. clinopodioides, CdNPs, and Daunorubicin. The charts display data with bars labeled with letters indicating statistical significance.
FIGURE 9

Leukemic myeloblasts infiltrating (%)

Spleen

Liver

Treatment

Control
Untreated
Cd(NO3)2
Z. clinopodioides
CdNPs
Daunorubicin

FIGURE 9
FIGURE 10

**Weight (mg)**

- Liver: Treatment effect
- Spleen: Treatment effect

**Volume (mm³)**

- Liver: Treatment effect
- Spleen: Treatment effect
FIGURE 11

(A) Results for different tissue samples under various treatments. The y-axis represents volume (mm³) and the x-axis represents different treatment categories.

(B) Similar analysis as in (A) but for different tissue samples with a different y-axis scale for volume (mm³).
<table>
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<tr>
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<td>10.70</td>
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<td>S</td>
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FIGURE 15

(a) and (b) SEM images of the sample surface. (c) and (d) Enlarged views with measured diameters:

- D3 = 36.36 nm
- D4 = 34.56 nm
- D1 = 20.08 nm
- D8 = 33.31 nm
- D6 = 20.33 nm
- D7 = 27.11 nm
- D2 = 27.72 nm
- D5 = 14.76 nm