Neuroprotective effects of erythropoietin against oxidant injury following brain irradiation: an experimental study

Gamze Ugurluer¹, Aysegul Cebi², Handan Mert³, Nihat Mert³, Meltem Serin¹, Haldun Sukru Erkal⁴

¹Department of Radiation Oncology, Faculty of Medicine, Acibadem University, Acibadem Adana Hospital, Adana, Turkey

²Department of Food Engineering, Faculty of Engineering, Giresun University, Giresun, Turkey

³Department of Biochemistry, Faculty of Veterinary, Yuzuncu Yil University, Van, Turkey ⁴Department of Radiation Oncology, Faculty of Medicine, Sakarya University, Sakarya, Turkey

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Abstract

Introduction: Radiation therapy (RT) is a major treatment modality, and the central nervous system is a dose-limiting organ in clinical RT. This experimental study aims to present the evaluation of the neuroprotective effects of erythropoietin (EPO) against oxidant injury following brain irradiation in rats.

Material and methods: Forty Wistar rats were randomly assigned to four groups (n = 10 each). In group 1 the rats received no EPO and underwent sham RT. The rats in groups 2 and 3 received EPO. In group 2 rats underwent sham RT, while in group 3 rats received RT. The rats in group 4 received no EPO and underwent RT. Rats were irradiated using a Cobalt-60 teletherapy machine using a single fraction of 20 Gy covering the whole brain. Cervical dislocation euthanasia was performed. The nitrite and malondialdehyde (MDA) levels and the superoxide dismutase (SOD) and glutathione peroxidase (GSHPX) activities were evaluated in dissected brain tissues.

Results: The nitrite and MDA levels were higher in the RT group (2.10 ±0.62 ppm, 26.02 ±2.16 nmol/ml; p < 0.05) and lower in the EPO + RT group (1.45 ±0.12 ppm, 25.49 ±1.90 nmol/ml; p < 0.05). The SOD and GSHPX activity was higher in the EPO + RT group (2.62 ±0.49 U/mg, 1.75 ±0.25 U/mg, p < 0.05). **Conclusions:** This study supports the probable neuroprotective effects of EPO against oxidant injury following brain irradiation in a rat model, presumably through decreasing free radical production and increasing expression of antioxidant enzymes.

Key words: brain, irradiation, oxidant injury, neuroprotective effects, erythropoietin.

Introduction

The central nervous system (CNS) is a dose-limiting organ in clinical radiation therapy (RT) [1]. Irradiation causes injury to the cellular DNA content, and this is most commonly a result of the reactive oxygen/nitrogen oxide species (ROS/RNOS) [2, 3]. The ROS/RNOS include oxygen ions, free radicals and both inorganic and organic peroxides (i.e., the su-

Corresponding author:

Gamze Ugurluer MD Department of Radiation Oncology Faculty of Medicine Acibadem University Acibadem Adana Hospital Cumhuriyet Caddesi No 66 01130 Adana, Turkey Phone: +90 5322509172 Fax: +90 3224554477 E-mail: gamze.ugurluer@ acibadem.edu.tr



peroxide radical (O_{2}) , the hydroxyl radical (OH), nitric oxide (NO) and hydrogen peroxide (H_2O_2)), which are highly reactive due to the presence of unpaired electrons in their shells [4]. Nitric oxide is a free radical induced by irradiation and has cytotoxic effects [5]. Due to the very short half-life of NO in aqueous solutions, it is generally measured indirectly via its metabolites (nitrate and nitrite), collectively referred to as reactive nitrogen intermediates. The hydroxyl radical reacts with extremely high rate constants with almost every type of molecule found in living cells: sugars, amino acids, phospholipids, DNA bases and organic acids [6]. Mechanisms have evolved to restrict and control such processes, partly by compartmentation, and partly by antioxidant defenses such as chain-breaking antioxidant compounds capable of forming stable free radicals and the evolution of enzyme systems (e.g. superoxide dismutase (SOD), catalase (CAT), and glutathione peroxidase (GSH-PX)) that diminish the intracellular concentration of the ROS [7]. Malondialdehyde (MDA), a naturally occurring product of lipid peroxidation and prostaglandin biosynthesis, is one of the most frequently used indicators of lipid peroxidation [8]. However, during times of environmental stress, the ROS levels can increase dramatically, overwhelming antioxidant systems and resulting in significant damage to cell structures [3]. This scenario is referred to as oxidative stress. Under normal conditions, ROS/ RNOS generation is approximately in balance with the cell's antioxidant defenses (antioxidants/antioxidant enzymes). Any imbalance between ROS/ RNOS generation and destruction in favor of ROS/ RNOS generation can create oxidative stress. The brain has relatively modest antioxidant systems [9]. Cancer patients, with many ongoing improvements in health care, survive long enough for the adverse radiation effects to become clinically significant and affect quality of life. Therefore, the development of any neuroprotective treatment that reduces radiation-related side effects is clearly desirable.

Erythropoietin (EPO) is a cytokine produced in the kidney that is predominantly involved in the proliferation, maturation and differentiation of the precursors of the erythroid cells. Human recombinant EPO has been used for over a decade to treat anemia in uremic as well as cancer patients [1]. Erythropoietin and its receptor, EpoR, have been shown to be expressed in rodent and human brains, in cultured neurons, astrocytes, oligodendrocytes, microglia, and endothelial cells, which have led to studies investigating the additional biological roles of EPO [10]. Erythropoietin and EpoR in the CNS and the upregulation of EPO by hypoxia/ischemia in vitro and in vivo suggest that this cytokine is an important mediator of the response of the brain to injury [11]. The suggested neuroprotective effects of EPO include antagonism against the deleterious action of glutamate, increase in the expression of antioxidant enzymes, decrease in free-radical production, augmentation in the release of neurotransmitters, increased pro-angiogenic activity and induction of neuroglobin (an intracerebral O, transporter) [12]. Despite its large molecular size (37 kD) and its highly glycosylated and negatively charged structure, systemic EPO crosses the blood-brain barrier in a dose-dependent manner to peak in the brain at 6 to 10 h [13]. Based on previously published animal studies, EPO is neuroprotective at 5000 IU/ kg [13]. Although the neuroprotective effects of EPO have been well documented in experimental models of brain and spinal cord injury, this experimental study aims to present the evaluation of the neuroprotective effects of EPO against oxidant injury following brain irradiation.

Material and methods

Experimental design

The experimental study procedures followed the "Guide for the Care and Use of Laboratory Animals (Canadian Council on Animal Care)". It was done at the "Inonu University Experimental Studies Laboratory". The protocol approval was obtained from "Inonu University Animal Ethics Committee". Forty female Wistar rats were randomly assigned to four groups (n = 10 each). They were four months old and weighed between 160 and 180 g. The rats were individually housed in rooms with controlled temperature and light. Standard chow and tap water were freely available. Group 1 was the control group and the rats did not receive EPO and underwent sham RT. Erythropoietin was given to the rats in groups 2 and 3. In group 2 rats underwent sham RT (EPO group), while in group 3 rats received RT (EPO + RT group). Group 4 was the RT group and the rats did not receive EPO and received RT. Erythropoietin (NeoRecormon, Roche, Mannheim, Germany) was prepared with dilution in 0.9% sodium chloride. It was administered (5000 IU/kg body weight) by intraperitoneal injection 8 h prior to RT or sham RT. For the rats in control and RT groups, the same volume (0.9%) of sodium chloride was administered by intraperitoneal injection.

Irradiation and euthanasia

The rats were irradiated in a prone position using a Cobalt-60 teletherapy machine. The fraction dose was 20 Gy and a single anterior portal with a 0.5 cm bolus covered the whole brain. The dose was defined at a depth of 2.5 cm. Ketamine (Ketalar, Pfizer, Istanbul, Turkey, 80 mg/kg body weight) and xylazine (Rompun, Bayer, Istanbul, Turkey, 5 mg/kg body weight) anesthesia was used and administered by intraperitoneal injection before irradiation. A customized restrainer was used for immobilization. Sham RT was also given on a Cobalt-60 teletherapy machine using the same duration. The rats were observed until they were completely recovered from the anesthesia. The cervical dislocation euthanasia was performed at 2 h following RT or sham RT.

Biochemical evaluation

The brains were dissected and frozen in crushed dry ice. The tissues were stored at -35° C until biochemical evaluation. Cold phosphate buffered saline (PBS) solution (50 mM, pH 7.4, with a ratio of 1 : 5) was used for perfusion to prevent blood contamination before tissue homogenization. A homogenizer (Yellow line DI-25 Basic, IKA Werke GmbH, Staufen, Germany) was used and the temperature was kept at +4°C during the process.

Total nitrate and nitrite concentrations were measured as the stable end product, nitrite, according to Stahr's modification method and determined spectrophotometrically at 520 nm [14]. For measurement of MDA level, homogenate and phosphoric acid were mixed and then heated for 45 min after the addition of thiobarbituric acid [15]. The absorbance of the complex was measured on a spectrophotometer (Boeco S-22 UV/ Vis, Hamburg, Germany).

The homogenate was centrifuged for 15 min using a centrifuge (Heraeus Sepatech Minifuge RF, Heraeus Instruments GmbH, Hanau, Germany). The assay for SOD activity was made according to the method of Suttle using a Ransel kit (Randox Labs., Crumlin, UK) [16]. This method is based on the formation of red formazan from the reaction of 2-(4-iodophenyl)-3-(4-nitrophenol)-5-phenyltetrazolium chloride and superoxide radical (produced in the incubation medium from xanthine oxidase reaction), which is assayed in a spectrophotometer at 505 nm. The inhibition of the produced chromogen is proportional to the activity of the SOD present in the



Figure 1. Nitrite levels by groups (group 1: control group, group 2: EPO group, group 3: EPO + RT group, group 4: RT group)

sample. 50% inhibition is defined as 1 unit of SOD (specific activity as units per milligram of protein).

The GSHPX activity was determined using the Ransel kit according to Paglia and Valentine [17]. GSHPX catalyzes the oxidation of glutathione by cumene hydroperoxide. In the presence of glutathione reductase and NADPH, the oxidized glutathione was immediately converted to the reduced form with concomitant oxidation of NADPH to NADP+. The decrease in absorbance was monitored with a spectrophotometer at 340 nm. One GSHPX unit is defined as 1 μ mol of NADPH consumed per minute, and specific activity is reported as units per milligram of protein.

Statistical analysis

The results were expressed as means \pm standard deviation (SD) for the nitrite and MDA levels and SOD and GSHPX activities. Student's *t* test was used for comparison. Statistical analysis was performed using SPSS (version 17.0), and *p* < 0.05 was considered as significant.

Results

The nitrite levels were 1.69 ±0.14 ppm (range: 1.4–1.87) for the control group, 1.66 ±0.23 ppm (range: 1.47–2.13) for the EPO group, 1.45 ±0.12 ppm (range: 1.27–1.60) for the EPO + RT group and 2.10 ±0.62 ppm (range: 1.33–3.20) for the RT group, and the difference was statistically significant (p < 0.05). There was a significant difference in nitrite levels among the control group and RT and EPO + RT groups; the nitrite level was higher in the RT group and lower in the EPO + RT group (p < 0.05) (Figure 1).

The MDA levels were 21.95 ±2.54 nmol/ml (range: 17.16–26.62) for the control group, 25.07 ±3.89 nmol/ml (range: 18.26–33.22) for the EPO group, 25.49 ±1.90 nmol/ml (range: 22.60–27.78) for the EPO + RT group and 26.02 ±2.16 nmol/ml (range: 23.19–29.27) for the RT group, and the difference was statistically significant (p < 0.05). There was a significant difference in MDA levels between the control group and the RT group; the MDA level was higher in the RT group (p < 0.05) and lower in the EPO + RT group (Figure 2).

The SOD activity was 1.73 ±0.13 U/mg (range: 1.54–1.94) for the control group, 1.95 ±0.15 U/mg (range: 1.72–2.14) for the EPO group, 2.62 ±0.49 U/mg (range: 1.80–3.15) for the EPO + RT group and 1.90 ±0.17 U/mg (range: 1.62–2.14) for the RT group, and the difference was statistically significant (p < 0.05). There was a significant difference in SOD activity between the RT and EPO + RT groups; the SOD activity was higher in the EPO + RT group (p < 0.05) (Figure 3).

The GSHPX activity was 1.67 ±0.16 U/mg (range: 1.33–1.86) for the control group, 1.54



Figure 2. MDA levels by groups (group 1: control group, group 2: EPO group, group 3: EPO + RT group, group 4: RT group)

±0.18 U/mg (range: 1.12–1.75) for the EPO group, 1.75 ±0.25 U/mg (range: 1.37–1.99) for the EPO + RT group and 1.71 ±0.29 U/mg (range: 1.35–2.13) for the RT group, and the difference was statistically significant (p < 0.05). There was increased activity in the EPO + RT group (Figure 4).

The relevant biochemical results are summarized in Table I.

Discussion

Radiation therapy is a major treatment modality that is received by about 60% of all cancer patients either for cure or palliation. The effectiveness of RT is limited by the potential for normal tissue injury. Radiation injury of the CNS has devastating clinical consequences. The associated pathologies are thought to be a consequence of complex interplay among many different processes including vascular endothelial cell loss, excessive generation of ROS and inflammatory mediators [18]. It is now recognized that clonogenic cell death is not the only mode of cell death in the CNS after RT [19, 20]. Certain glial, neuronal and endothelial cells in the CNS also undergo apoptosis soon after RT. Furthermore, there is a component of secondary injury and cell death that is mediated by microenvironmental alterations such as hypoxia/ischemia and inflammation [1]. Similar to other CNS injury models associated



Figure 3. SOD activity by groups (group 1: control group, group 2: EPO group, group 3: EPO + RT group, group 4: RT group)

with an increase in ROS and a state of oxidative stress, the same was observed in the irradiated CNS. Similar with the literature in our study, the nitrite and MDA levels were increased in the RT group as an indicator of lipid peroxidation and presence of radiation-induced oxidative damage. Studies comparing MDA levels in rat brains after irradiation or sham irradiation showed that MDA levels for the rats that underwent irradiation were significantly higher than for the rats that underwent sham irradiation or sham surgery [21, 22]. High levels of MDA are indicative of oxidative stress [23].

Strategies aimed at blocking the effector molecules or otherwise reducing oxidative stress are attractive for preventing or mitigating radiation-in-



Figure 4. GSHPX activity by groups (group 1: control group, group 2: EPO group, group 3: EPO + RT group, group 4: RT group)

Parameter Groups Control EPO EPO + radiation **Radiation therapy** therapy 1.69 ±0.14 Nitrite [ppm] 1.66 ±0.23 1.45 ±0.12 2.10 ±0.62 (1.40 - 1.87)(1.47 - 2.13)(1.27 - 1.60)(1.33 - 3.20)21.95 ±2.54 25.07 ±3.89 25.49 ±1.90 26.02 ±2.16 MDA [nmol/ml] (17.16 - 26.62)(18.26 - 33.22)(22.60 - 27.78)(23.19 - 29.27)SOD [U/mg] 1.73 ±0.13 1.95 ±0.15 2.62 ±0.49 1.90 ±0.17 (1.54 - 1.94)(1.72 - 2.14)(1.80 - 3.15)(1.62 - 2.14)GSHPX [U/mg] 1.71 ±0.29 1.67 ±0.16 1.54 ±0.18 1.75 ±0.25 (1.33 - 1.86)(1.37 - 1.99)(1.35 - 2.13)(1.12 - 1.75)

 Table I. Nitrite and MDA levels and SOD and GSHPX activities by groups

duced CNS toxicity. One approach to impeding the actions of reactive oxidants is to administer drugs that induce endogenous antioxidant biochemical processes.

Human recombinant EPO has been used clinically for more than two decades to treat anemia associated with conditions such as chronic renal disease, in antiviral HIV therapy, and in cancer patients on chemotherapy. In an attempt to augment oxygen delivery as a response to hypoxia, EPO exerts its action through the up-regulation of specific receptors that have been shown to be abundant not only in the bone marrow, but also in a variety of tissues including the brain. Besides the expression the EPO receptors in the brain, autocrine production of EPO has been reported in the astrocytes as well as the oligodendrocytes, the glial cells and the endothelial cells [24].

In studies concerning the clinical use of EPO to correct anemia in patients with chronic renal diseases and cancer, an emphasis has been placed on the improvement in the quality of life as well as the improvement in the cognitive function [12, 25, 26]. Consequently, the effects of EPO on the quality of life and the cognitive function were evaluated in a placebo-controlled randomized pilot study by O'Shaughnessy et al. on patients with breast cancer undergoing chemotherapy, and EPO was found to attenuate the cognitive impairment associated with chemotherapy [27]. These clinical observations have indicated a direct potential neuroprotective role of EPO, and numerous studies are in favor of a neuroprotective role of EPO.

There is now an emerging body of data on the neuroprotective effect of EPO against a wide variety of CNS insults [28]. Local production of endogenous EPO in the brain in a hypoxia-dependent manner further suggests that EPO may act in a paracrine or autocrine manner to provide neuroprotection. Analogous to its activity in erythroid progenitor cells, EPO can stimulate the survival and proliferation of neural progenitor cells. Erythropoietin neuroprotection in animal models of brain ischemia/trauma raise the possibility for EPO therapy in brain injury/disease [29].

Erythropoietin use in stroke patients is appealing since it is well studied. Intravenous high-dose EPO in a total of 53 stroke patients was well tolerated and associated with an improvement in clinical outcome without any signs of elevated hematocrit levels [30]. With such encouraging early results, EPO investigations are being expanded to include Parkinson's disease, HIV-mediated neuropathy, diabetic neuropathy, macular degeneration, retinitis pigmentosa, glaucoma, demyelinating injury, schizophrenia, and other types of neurodegenerative diseases [31, 32]. Therefore, a potential neuroprotective role has been advocated for EPO in the setting of a variety of conditions affecting the central nervous system.

The suspected neuroprotective effects of EPO include a decrease in ROS generation, an increase in the expression of antioxidant enzymes, augmentation of the release of neurotransmitters, increased pro-angiogenic activity and the induction of neuroglobin (an intracerebral O transporter), hence antagonizing the deleterious action of glutamate, the inhibition of apoptosis [12]. This vast array of protective properties, most likely related to the prominent role of EPO during brain development, may explain the potent neuroprotective action of EPO in various different animal models. These, mostly rodent, models range from hypoxia/ischemia to traumatic injury of the brain or spinal cord, subarachnoid hemorrhage, multiple sclerosis, Parkinson's disease, epilepsy, retina degeneration, neuropathies including diabetes, radiation-induced brain damage, and cerebral malaria, to mention just the most obvious examples [33].

Erythropoietin may also decrease NO-mediated injury, but it does not affect NO levels directly. The mechanisms involve reduced lipid peroxidation and increased glutathione peroxidase [34, 35]. In our study, the nitrite and MDA levels were lower in the EPO group than in the RT group. The SOD and GSH-PX activity was higher in the EPO-treated group.

Despite a large molecular size and a highly glycosylated and negatively charged structure, EPO has been shown to cross the blood-brain barrier in a dose-dependent manner [13]. After a single high-dose intravenous injection, EPO is detectable in the brain within hours, reaching a peak concentration at 3–4 hours in the brain or cerebrospinal fluid in humans and other animals [24].

Based on experimental studies, EPO is neuroprotective starting at a dose of 5000 IU/kg [13]. Thus, following a high EPO dose (5000 U/kg *i.v.*) there was a peak in the CSF 4 h after injection, and a significant correlation was observed between blood and CSF EPO concentrations [12]. Thus, we used this dose in our study.

In conclusion, this study supports the probable neuroprotective effects of erythropoietin against oxidant injury following brain irradiation in a rat model, presumably through decreasing free radical production and increasing expression of antioxidant enzymes. Nevertheless, the use of EPO as a neuroprotectant for radiation-induced oxidant injury warrants further evaluation regarding both effectiveness and safety.

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Conflict of interest

The authors declare no conflict of interest.

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